# Phosphorus in industry and society

Willem Schipper



### world usage of phosphorus



### fertilizers

- main use of phosphorus worldwide
- P indispensible to life
- energy
- DNA
- bones

### DNA



### your body gets its energy here







with phosphate

> without phosphate

### Fertilizers



### most commonly used fertlizer ingredients (P)

- phosphoric acid (MGA)
- MAP
- DAP
- TSP
- (SSP)



### feed phosphates

- animal husbandry
- animals need P to exist AND grow
- P in fodder not sufficient and not all digestible
- feed additives needed
- MCP, DCP, mixtures
- about 5% of P use



linked to manure issue

phytic acid (largely non-digestible but rich in P)



### value chain for phosphorus



### P<sub>4</sub> - the key to P chemistry

- made in submerged arc furnace process
- not unlike a blast furnace
  - submerged arc furnace

### the phosphorus production process





Tennessee Valley Authority P furnace, 1942





#### Phosphorus family tree

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### uses of $P_4$ as such

limited:

- military (incendiaries, e.g Hamburg 1943)
- smokescreens
- rat poison (obsolete)
- homeopathy



Emsley J; The 13th Element: The Sordid Tale of Murder, Fire and Phosphorus; Wiley and Sons: New York, **2000** 

### glyphosate

 $P_4 \rightarrow PCI_3$ PCI<sub>3</sub> + IDA + CH<sub>2</sub>O -> glyphosate





most sold herbicide worldwide

blocks 5-enolpyruvylshikimate-3-phosphate (EPSP) synthase (shikimate pathway to amino acids)

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US EPA 2000–2001 Pesticide Market Estimates





concerns about brominated flame retardants





phosphate based alternatives

Weil, E.D.; Levchik, S.V.; Flame Retardants for Plastics and Textiles; Hanser, München 2009

### phosphonates

- P analogs of EDTA etc
- chelating agents
- industrial water treatment, RO, detergents



### Li ion batteries

- P compounds in battery
- conductivity
- Li storage





LiPF<sub>6</sub> and LiFePO<sub>4</sub>



## engine oil additive ZDDP





increases wear resistivity; antioxidant

 $P_2S_5$  + iPr-OH + ZnO -> Zinc O,O'-diisopropyl dithiophosphate

Ρ

### electroless nickel plating

 $P_4$  + NaOH -> NaH<sub>2</sub>PO<sub>2</sub>

NaH<sub>2</sub>PO<sub>2</sub> + Ni salt -> Ni layer

Actually Ni / Ni<sub>3</sub>P



A. Wurtz; Compt. Rendus Acad. Sc. 1844, 18, 702

### FOOD



solid acid for baking powder



emulsifying agent





moisture retention in cooking







acidulation

Na, K, Horthophosphates pyrophosphates tripolyphosphates *blends* 

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### metal extraction

separation of cobalt from nickel





most of the world's cobalt is produced by using this chemical



### catalysis

phosphine chemicals are essential parts of catalysts

e.g. phobane in hydroformylation

major pathways in petrochemistry



phobane







### osteoporosis drugs







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### drug synthesis



ring closure for the production of APIs



Quetiapine

treatment for schizophrenia and depression

### asphalt additive



polyphosphoric acid

to improve asphalt characteristics under extreme circumstances



### Closing the loop

P stewardship in technosphere (P<sub>4</sub>) and biosphere not independent

Recycle streams into technosphere P (white phosphorus ----- P rock replacement)



### Phosphorus product routes

surprisingly, P<sub>4</sub> derivatives P ends up in five broad categories only



### white phosphorus – sustainable?

- P essential use = agriculture, it should perhaps be "reserved" for that?
- however only 2% of P is used in true P<sub>4</sub> chemistry
- 2% more P use efficiency can probably be achieved more easily in agricultural applciation improvements and reducing meat consumption (feed phosphate – 5 to 10% of total)
- consider cradle-to-cradle design for P chemicals (e.g. recycle flame retarded plastics as such) – leading C2C principle: use freely and reuse, instead of abandon uses and forbid applications, therefore a limit on non-agricultural uses of P does not seem needed at this point

### Thank you for your attention!

#### Willem Schipper

Consulting for phosphorus, P derivatives, phosphoric acid and phosphates

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