

Magpie

Polymers for Precious Metal Capture & Recovery

Metal Scavenging:

using low-value phosphorus materials to make metal refining more sustainable

http://www.magpie-polymers.com

Steven van Zutphen December 2016

European Sustainable Phosphorus Platform













Selective capture precious metals

Filtration of **industrial effluents**

Recover value from waste































EFFICIENT RECYCLING KEY FOR SUSTAINABILITY



Sources: Sustainable Development Report 2009, p56, Anglo Platinum Sustainability Reporting and the Platinum Group Metals, G.M. Mud 2012, Nature Biotech 1999, p541 PGM recycling at Hereaus, 2012



Manufacturing of **high-value chemicals**

Innovation driven by economics

Real and measurable impact on health, safety and environment



Founded in 2011

450 m² facilities **1h** from Paris

10 people, **10** products, **10** countries

Based on innvation from 2007



Magpie

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Discovery of something exisiting but unknown

Invention of something new

Innovation by combining things in a novel way



MAGPIE USES PHOSPHORUS UNLIKE THE OTHERS



Typical ion-exchangers

- Use acids (sulphonate or carboxylic) for cation exchange
- Use amines (tertiary and quaternairy) for anion exchange
- Use sulfur (thiol, thiouronium) for chelating functional groups

Magpie uses phosphorus

- Unique coordination properties of phosphines, phosphine oxide and phosphonates
- No need for PCl or PLi type species in production
- Use PH bonds for our chemistry



PRODUCT AND PROCESS INNOVATION



Process development example:

- Surface treatment industry
- Use our knowledge in solution chemistry

Product development example:

- Silver refining industry
- Exploit stability of our materials



PLATING ON PLASTIC



POP, complex multistep process

- Layer decorative or functional gold/nickel/copper onto ABS plastic
- Key-step involves Pd(0) catalyst reducing a thin layer of metal on the surface
- Catalyst applied in Pd/Sn colloid form and reduced in-situ







PROBLEM FACED





MAGPIE SOLUTION

Colloid is destablized while retaining palladium in **Pd(II)** form

Avoid **co-precipitation** of palladium and tin together

Scavange the palladium onto phosphine-oxide resin MPS-1207

Leave most tin in waste water to be removed at the water treatment station (hydroxide precipitation)



Step1:	Palladium Loading
1% volume added, 15 min	25.7 g/L of MPS-1207



INDUSTRIAL SOLUTION



Automated system:

- Carries out the two step process with minimum intervention
- Recovers palladium on columns that can be changed when saturated



SILVER MARKET DEMANDS INCREASED PURITY





<3N Silver



- 50/50 industrial / non-industrial uses of silver
- Electronics grade requires highest purity
- up to 30% higher in price
- 0.01% of impurity or 100 mg/kg





Step 1: obtain 80-90% pure silver







Step 2: obtain 99.9-99.99% pure silver



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"THE REASONABLE MAN ADAPTS HIMSELF TO THE WORLD; THE UNREASONABLE ONE PERSISTS IN TRYING TO ADAPT THE WORLD TO HIMSELF. THEREFORE ALL PROGRESS DEPENDS ON THE UNREASONABLE MAN."

GEORGE BERNARD SHAW

C Lifehack Quotes













